



MATERIAL SAFETY DATA SHEET

CHEMICAL IDENTITY Sulfuric acid

1. Product and Company Identification

PRODUCT NAME: Stop Solution: a component of below products

HBs S Antigen Quantitative ELISA Kit, Rapid-II (BCL-SHP-2), HB Pre-S1 Antigen Quantitative ELISA Kit, Rapid (BCL-S1HP-01), HB Pre-S2 Antigen Quantitative ELISA Kit, Rapid (BCL-S1HP-01), Easy ELISA constructor (BCL-EEC-01), Leishmaniasis antibody detection ELISA kit (BCL LEI-01), Filariasis antibody

detection ELISA kit (BCL FIL-01)

BCL-SS2, BCL-SS3, BCL-ESS-01 PRODUCT CODE:

CHEMICAL IDENTIFICATION: Sulfuric acid CONTENT IN THE PRODUCT: 9.80%

RECOMMENDED USE: HBsAg S, Pre-S1 and Pre-S2 antigen detection by ELISA (for research use only)

ELISA construction for detecting antigen-specific antibody (for research use only) Leishmania and Filariasis antibody detection of Urine by ELISA (for research use only)

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(THE BELOW DESCRIPTION IS APPLIED FOR THE IDENTIFIED CHEMICAL OF 95%)

2. Hazards Identification

GHS hazard class and category PHYSICAL HAZARDS **EXPLOSIVES**

FLAMMABLE GASES FLAMMABLE AEROSOLS OXIDIZING GASES GASES UNDER PRESSURE FLAMMABLE LIQUIDS

FLAMMABLE SOLIDS SELF-REACTIVE SUBSTANCES AND MIXTURES PYROPHORIC LIQUIDS PYROPHORIC SOLIDS SELF-HEATING SUBSTANCES AND MIXTURES

SUBSTANCES AND MIXTURES WHICH, IN OXIDIZING LIQUIDS OXIDIZING SOLIDS ORGANIC PEROXIDES CORROSIVE TO METALS

ACUTE TOXICITY ORAL
ACUTE TOXICITY DERMAL HEALTH HAZARDS 5

ACUTE TOXICITY - INHALATION (GAS) ACUTE TOXICITY - INHALATION (VAPOURS) ACUTE TOXICITY - INHALATION (DUST/MIST) SKIN CORROSION/IRRITATION EYE DAMAGE/IRRITATION SENSITIZATION RESPIRATORY SENSITIZATION SKIN

GERM CELL MUTAGENICITY CARCINOGENICITY TOXIC TO REPRODUCTION

SPECIFIC TARGET ORGAN SYSTEMIC TOXICITY 1 (respiratory system) SPECIFIC TARGET ORGAN SYSTEMIC TOXICITY 1 (respiratory system)

ASPIRATION TOXICITY ENVIRONMENTAL HAZARD HAZARDOUS TO THE AQUATIC ENVIRONMENT-HAZARDOUS TO THE AQUATIC ENVIRONMENT -

LABEL ELEMENTS

PREVENTTION

SYMBOL

SIGNAL WORD

HAZARD STATEMENTS May be harmful if swallowed

Fatal if inhaled

Causes severe skin burns and eye damage

Causes serious eye damage

Causes damage to organs (respiratory system)

Causes damage to organs through prolonged or repeated exposure (.respiratory system)

Harmful to aquatic life

Do not handle until all safety precautions have been read and understood.

Do not eat, drink or smoke when using this product

Wear respiratory protection.

Do not breathe dust/fume/gas/mist/vapours/spray. Avoid contact during pregnancy/while nursing.
Use personal protective equipment as required. Use only outdoors or in a well-ventilated area. Wash hands thoroughly after handling

Avoid release to the environment.

Reponse IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to

If eye irritation persists, get medical advice/attention.

IF INHALED: Remove victim to fresh air and keep at rest in a position comfortable for breathing. IF ON SKIN: Wash with plenty of soap and water. If skin irritation occurs, seek medical advice/attention.

Call a POISON CENTER or doctor/physician if you feel unwell.

Store locked up. Storage

Store container tightly closed in cool/well-ventilated place DISPOSAL Dispose of contents and container in accordance with regulation. 3. Composition/Information on Ingredients

Substance/Preparation Preparation Chemical name Sulfuric Acid

Synonyms Oil of vitriol, Hydrogen sulfate

Ingredient name Sulfuric Acid Composition(%) 9.80% H2SO4(98.08) Chemical formula(MW)

Structural Formula

7664-93-9 CAS No. (1)-430 Chemical No. Japan

4. First-Aid Measures

Inhalation

Ingestion

Skin contact

Move victim to fresh air General procedures

Keep victim warm and quiet. Call emergency medical service

Apply artificial respiration if victim is not breathing Administer oxygen if breathing is difficult.

Do not use mouth-to-mouth method if victim ingested or inhaled the substance; induce artificial respiration

Remove and isolate contaminated clothing and shoes.

In case of contact with substance, immediately flush skin or eyes with running water for at least 20 Ensure that medical personnel are aware of the material(s) involved, and take precautions to protect

In case of accident by inhalation: remove casualty to fresh air and keep at rest.

If breathing is week, irregular or has stopped, open his airway, loosen his collar and belt and administer

Removal of solidified molten material from skin requires medical assistance.

For minor skin contact, avoid spreading material on unaffected skin. Take off immediately all contaminated clothing.

After contact with skin, wash immediately with plenty of water.

Remove all chemicals from contact with the victim ≠s skin as quickly as possible. A delay of only seconds Eye contact

In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. Remove all chemicals from contact with the victim's eyes immediately. A delay of only seconds may

If swallowed, do not induce vomiting: seek medical advice immediately and show this container or label.

If swallowed, rinse mouth with water (only if the person is conscious). Immediately give the person one or two glasses of milk or water, to dilute the chemical, do not attempt to

Protect yourself by wearing rubber gloves and air-tight safety goggles. Effects of exposure (inhalation, ingestion or skin contact) to substance may be delayed Precaution for first-aid persons

Information for the physician

5. Fire-Fighting Measures

The product is non-flammable. In case of fire, use dry powder CO2 dry sand. Extinguishing Media

Incompatible extinguishing media Never use water. When material is not involved in fire: do not use water on material itself. Specific Hazards with regard to Fire-Fighting

Fire may produce irritating, corrosive and/or toxic gases

Runoff from fire control or dilution water may cause pollution.

Toxic gases will form upon combustion of: sulfur oxides
This product is noncombustible. When surrounding fire, move containers from fire area if it can be done
without risk, if not possible, apply water from a safe distance to cool and protect surrounding area. Dry
chemical powder, carbon dioxide or dry sand should be used for small fires. Specific fire-fighting measures

Protection of fire-fighters Firefighters should wear proper protective equipment

6. Accidental Release Measures

Measures for Handling Personnel Do not touch damaged containers or spilled material unless wearing appropriate protective clothing.

Evacuate non essential personnel Wear proper protective equipment.

Measures for environmental effects Attention should be given not to cause damage to the environment by flowing of spillage to rivers.

Measures when handling spilled substances Shut off the leakage source to stop the leakage. For small spill, absorb spill with absorbent and move to a

Preventive measures for secondary accident Keep unauthorized personnel away.

Keep combustibles (wood, paper, oil, etc.) away from spilled material.

Use water spray to reduce vapors; do not put water directly on leak, spill area or inside container.

7. Handling and Storage

Handling

(Exposure Control for handling personel) Wear proper equipment not to contact with skin or inhale the Contact with molten substance may cause severe burns to skin and eyes. Preventive measures Safety treatments

Use with an enclosed system or a local exhaust ventilation.

Safety Measures/Incompatibility Wear proper equipment not to inhale and contact with skin, eyes and clothes.

Do not shock, overturn, drop, or drag containers.

Storage

Incompatible materials Corrodes many kinds of metals Recommendation for storage Keep tightly closed in dark cool place Incompatible storage condition Keep away from organic substances

Keep away from combustible materials Recommendation on container and Glass, Teflon, polyvinyl chloride resin, etc.

packaging materials

8. Exposure Control/Personal Protection Control value Not established

Adopted value (Sulfuric acid contained in strong inorganic acid mists)

ACGIH 2005 TLV: 0.2 mg/m3 Thoracic fraction (Sulfuric acid)

(ceiling limit) 1 mg/m3 (under study) JCGIH 2005 TWA 1 mg/m3 TWA 1 mg/m3 NIOSH REL OSHA PEL

MAK (Inhalable fraction) 0.1 mg/m3 Peak limitation category: I(1)

Carcinogen category: 4 Pregnancy risk group: C

Engineering measures Keep source tightly closed or install local exhaust ventilation. Provide shower and vanity unit nearby and make clear the location of these

Personal protective equipment Wear positive pressure self-contained breathing apparatus (SCBA). Respirator for acidic vapor Respiratory protection

Wear impervious glove made from chloroprene, as appropriate Hand protection Wear protective eyeglasses or chemical safety goggles. Eye protection

Skin and body protection To prevent any contact, wear impervious clothing such as apron, boots, or whole-body suits made from

9. Physical and Chemical Properties

Physical properties Oily liquid Colorless Appearance Color Odor

None рΗ Melting point

Strong acid -32.0 C (93.10%), -16.5 C (95.05%), +3.0 C (98.00%) -29.4 C (93.19%), -22.2 C (95.00%), -1.1 C (98.00%) Freezing point Boiling point 279 C (93.19%), 297 C (95.00%), 327 C (98.00%)

Flash point Ignition temperature Explosion

0.13 kPa (146 C) Vapor pressure Relative Vapor Density (Air=1) 3.4

Specific gravity/Density About 1.84 Solubility Solubility in water: Miscible

Octanol /water partition coefficient (log Temperature of decomposition

Viscosity

Other data Hygroscopic

10. Stability and Reactivity Stability

This product is considered a stable material under normal and anticipated storage and handling conditions.

Reactivity Generates heat when contacted with water.

Sunlight, heat, contact with alkaline substances and combustible materials Conditions to avoid

Hazardous decomposition products Toxic fumes of Sulfur oxides

11. Toxicological Information Acute toxicity

Labor standard law, Japan

Emargency response guidebook Oral, Dermal, Inhalation Toxicities

TOXIC; inhalation, ingestion or contact (skin, eyes) with vapors, dusts or substance may cause severe

rat LD50 2140 mg/kg mouse LC50 320 mg/m3/2H Oral Inhalation mouse rat LC50 510 mg/m3/2H human TCL0 1 mg/m3/3H

Risks of serious damage to eyes. Causes severe burns. (1A-1C) Irritant propertie

250 ug ; SEVERE (1) Irritant properties to skin/eye 5 mg/30S rinse; SEVERE

Carcinogenic effects

(Sulfuric acid contained in strong inorganic acid mists) ACGIH-A2: Suspected human carcinogen. (Strong Inorganic Acid Mists Containing Sulfuric Acid) IARC-Gr.1: Carcinogenic to humans. (Strong Inorganic Acid Mists Containing Sulfuric Acid) NPT-Gr.a: Known to be Human Carcinogens.

Toxicity for reproduction Allergenic and sensitizing effects

Chronic toxicity Prolonged inhalation may cause upper respiratory infection, bronchitis, teeth acid erosion.

Mutagenic effects Cytogenetic analysis: hamster ovary 4 mmol/l

Teratogenic effects Narcotics and Psychotropics Control Law, Raw PsychoDrug, >20kg

12. Ecological Information

Biotranspotability Persistence and degradability Bioaccumulative potential

Ecotoxicity Harmful to aquatic organisms(3) Fish toxicity LD50/96H: 100-10 mg/l The lethal dose of fish: 6.3 mg/I/24H

13. Disposal Considerations

Contact a licensed professional waste disposal service to dispose of this material. Comply with all country,

national and local regulations. Do not dump this product into sewers, on the ground or into any body of

14. Transport Information UN No., UN CLASS

UNNo. 1830 **UN CLASS** 8 PG

Proper shipping name ER2000 GUIDE NO SULPHURIC ACID

F137

Protect from direct sun light in transportation, and confirm the container does not leak. Carefully load it Specific safety measures and conditions on

15. Other information

The information described above is believed to be correct to be the best of our knowledge and information but does not purport to be all inclusive and shall be used only as a guide. The product should be used by expert persons having knowledge and skill with their own risk. Beacle shall not be held liable for any damageresulting from handling or from contact with the above material.